

Readily Accessible, Modular, and Tuneable BINOL 3,3'-Perfluoroalkylsulfones: Highly Efficient Catalysts for Enantioselective In-Mediated Imine Allylation

Robert Kargbo,[†] Yoko Takahashi,[†] Santosh Bhor,[†] Gregory R. Cook,^{*,†} Guy C. Lloyd-Jones,^{*,‡} and Ian R. Shepperson[‡]

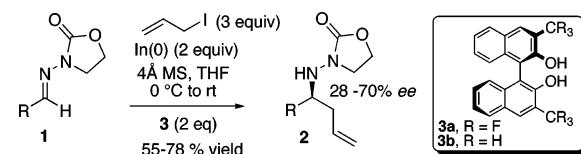
Department of Chemistry and Molecular Biology, North Dakota State University, Fargo, North Dakota 58105-5516, and Bristol Centre for Organometallic Catalysis, University of Bristol, Cantock's Close, Bristol, BS8 1TS, U.K.

Received February 1, 2007; E-mail: gregory.cook@ndsu.edu; guy.lloyd-jones@bris.ac.uk

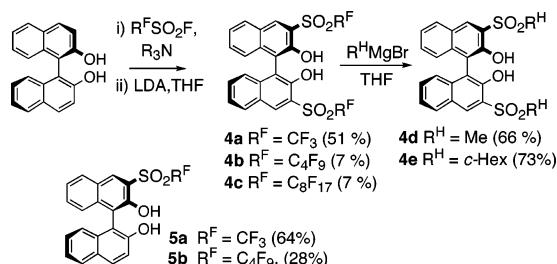
The development of methodologies for the enantioselective allylation of imines¹ is an area that has received much attention.² The Si-mediated allylation of *N*-acylhydrazones has been especially successful in this regard. For example, Leighton's modified allylsilane reagents³ and Kobayashi's sulfoxide/allyltrichlorosilane combination⁴ both provide good selectivity (up to 93% ee). Allylindium reagents⁵ in combination with chiral auxiliaries such as sulfonimine derivatives,⁶ amino acid-derived imines,⁷ and α -keto chiral sultams⁸ are also effective, and in a prior contribution to this area, we reported⁹ that *N*-acylhydrazones bearing an oxazolidinone auxiliary undergo In-mediated allylation with complete diastereocontrol.¹⁰ We subsequently developed an enantioselectively catalyzed allylation of *N*-acylhydrazones (**1** \rightarrow **2**, Scheme 1). An optimized BINOL derivative (**3a**) was found to induce 70–92% ee,^{11,12} and recently Jacobsen reported a urea-based catalyst giving the highest selectivity to date in an analogous process (76–95% ee, R = Ar; 80% ee, R = *i*Pr).¹³ Herein we report on modular and tuneable sulfone BINOLs (**4** and **5**) as readily accessible new ligands that prove outstanding in In-mediated allylation, *giving 2 in up to 99% yield and 99% ee*.

One key observation in our prior studies¹¹ was that the electronic, not steric, demands of the BINOLs **3** proved crucial in determining both the activity and selectivity. For example, the methyl derivative **3b** gave **2a** in 28% ee, inferior to the parent BINOL (\rightarrow 45% ee), itself less effective than the optimized 3,3'-bistrifluoromethyl system **3a** (\rightarrow 70% ee).¹¹ Although we have been unable, so far, to isolate the catalyst, NMR evidence (see Supporting Information) suggests that **3a** does not interact directly with the hydrazones (unlike the H-bonding urea catalysts of Jacobsen)¹³ but does react with the *in situ* generated allylindium.¹⁴ The latter process is associated with an up-field shift of the aromatic ¹H signals, consistent with BINOL deprotonation by the relatively nonbasic allylindium reagent. As the allylation reaction **1** \rightarrow (\pm)-**2** proceeds in the absence of BINOLs (R = Ph, 100% conversion in 11.5 h at room temp) improving catalyst activity is a prerequisite to higher enantioselectivity. On this basis, we sought more Brønsted acidic BINOLs leading to increased Lewis-acidity in the indium BINOLates. Sulfones in general,¹⁵ and perfluoroalkylsulfones (SO₂R^F) in particular, are significantly more electron withdrawing aromatic substituents than CF₃. Indeed, the Hammett σ_p value for SO₂CF₃ ('triflone') is 0.96 compared to a value of 0.54 for CF₃.¹⁵ We recently reported an LDA-mediated thia-Fries rearrangement route to ortho phenolic triflones,^{16a} subsequently made general by Butenschön.^{16b} We have now conducted double thia-Fries rearrangements which yield electron-demanding, enantioERICALLY pure 3,3'-bistriflone (**4a**), -nonaflone (**4b**), and -heptadecaflone (**4c**) systems, *in just two steps from BINOL*, Scheme 2. While the (unoptimized) yields of **4bc** are low, they represent the first examples of the rearrangement¹⁶ of higher perfluoroalkylsulfones.

Scheme 1. Enantioselective Allylation Using BINOLs (**3**)¹¹



Scheme 2. Rearrangement Route to New Sulfone Ligands



The sulfone BINOL ligand set is readily tuned. For example, mono-rearrangement affords SO₂R^F ligands (**5ab**), and the reaction of *unprotected* **4a** with excess R^HMgBr¹⁷ yields alkylsulfones **4d** and **4e**.¹⁸ These new 3- and 3,3'-sulfone BINOLs were evaluated at 10 mol % loading in the allylation of hydrazones **1ab**. Key results are shown in Table 1, where a stark contrast emerges between **3a** (entries 1 and 2) and the new SO₂R^F catalysts **4a–c** (entries 3–6).

As with **3ab**, use of THF as solvent, with 4 Å MS to scavenge water, is essential for selectivity.¹¹ However, the greater activity of catalysts **4a–c** was sufficient to facilitate a switch to an allyl bromide-derived indium reagent, affording even higher selectivity (compare entries 3 and 4). The presence of two SO₂R^F substituents was found to be crucial. The SO₂R^H catalysts **4d** and **4e** gave markedly lower selectivity, particularly for **2b** (entries 7 and 8) and intriguingly, the unsymmetrical catalyst **5a** afforded only 9–11% ee (entry 9). Bromo substituents in the 6,6' positions of **4a** and **5a** (from 6,6'-Br₂-BINOL) did not significantly alter the results. The remarkable efficacy of bis-SO₂R^F BINOL **4a** was also manifest in a more extensive study, Table 2.

Hydrazones possessing an ortho substituent gave particularly good results (96–99% ee, entries 6–15), facilitating lower catalyst loadings. Indeed, just 3 mol % of catalyst **4a** (entry 9) still gave **2f** in 98% ee. Aliphatic substrates are commonly problematic for the asymmetric allylation reaction, possibly because of their greater reactivity. However, **4a** again afforded significantly improved enantioselectivity (entry 17, 74% ee) as compared to the first generation catalyst **3a** (34% ee).¹¹ This represents the best example to date of catalytic enantioselective indium-mediated allylation of a linear¹³ aliphatic substrate.

In conclusion, we report modular and tuneable new enantioERICALLY pure sulfones, accessible in just two steps from BINOL.

[†] North Dakota State University.
[‡] University of Bristol.

Table 1. Comparison of BINOL **3a** with Sulfones **4/5** (10 mol %)^a

entry	BINOL (10 mol %)	2a (R = Ph)		2b (R = styryl)	
		yield %	ee % ^b	yield %	ee % ^b
1	3a	77	70	79	34
2	3a (200 mol %)	78	70		
3	4a	95	88	85	90
4 ^{c,d,e}	4a	82	90	87	97
5 ^c	4b	68 ^c	89 ^c		
6 ^c	4c	72 ^c	88 ^c		
7	4d	49	49	71	44
8	4e	98	68	87	10
9	5a	59	11	70	9

^a Conditions as Scheme 1. ^b Chiral HPLC. ^c Using allyl bromide. ^d 100% conversion, 4 h, 0–4 °C. Same conditions, with 0 mol % **4a**, 23% (\pm)-**2a**. ^e No reaction in DMF, CH₂Cl₂, or MeCN. In 75% aq THF **2a** obtained in 20% yield, 33% ee.

Table 2. Enantioselective Allylation Catalyzed by **4a** (10 mol %)^a

entry	hydrazone		X = I		X = Br	
			yield % ^b	ee % ^c	yield % ^b	ee % ^c
1	a		95	88	82	90
2	b		85	90	87	97
3	c		95	87	98	87
4	d		88	95	96	95
5	e		98	90	96	92
6	f		98	97	96	99
7			(4a 5 mol%)	99	99	
8			(4a 4 mol%)	96	98	
9			(4a 3 mol%)	94	98	
10			(4a 2 mol%)	94	95	
11			(4a 1 mol%)	95	88	
12	g		95	97	99	99
13	h		84	97	98	99
14	i		97	94	98	96
15	j		95	96	96	98
16	k		97	90	94	92
17	l		93	74	88	70

^a The allylindium/ligand complex was preformed. See Supporting Information for details. ^b Yield and ee are the average of at least two runs.

^c Chiral HPLC.

The bis-SO₂R^F BINOL **4a** facilitates a general and highly enantioselective catalytic indium-mediated imine allylation, affording **2** in up to 99% ee. This represents the highest selectivity to date in any indium-mediated allylation and the catalyst is easily recovered (silica-gel chromatography) and recycled ($\times 3$) without loss of activity or selectivity. The SO₂R^F BINOL systems (**4a–c**) offer significant opportunities for exploiting fluororous phase technologies¹⁹ in this and other reactions.

Acknowledgment. We are grateful to the National Science Foundation (Grant NSF-CHM-0616485) and the European Union (STRP 505167-1 LIGBANK) for financial support. We thank Pfizer and ND-EPSCoR for graduate fellowships (R.K.).

Supporting Information Available: Experimental procedures and characterization data for all new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

References

- Nucleophilic addition to imines: (a) Kobayashi, S.; Ishitani, H. *Chem. Rev.* **1999**, *99*, 1069. (b) Bloch, R. *Chem. Rev.* **1998**, *98*, 1407. (c) Enders, D.; Reinhold, U. *Tetrahedron: Asymmetry* **1997**, *8*, 1895. (d) Denmark, S. E.; Niclaise, O. J.-C. *Chem. Commun.* **1996**, 999.
- Recent examples of enantioselective imine allylation: (a) Fang, X.; Johannsen, M.; Yao, S.; Gatheroood, N.; Hazell, R. G.; Jørgensen, K. A. *J. Org. Chem.* **1999**, *64*, 4844. (b) Fernandes, R. A.; Stimač, A.; Yamamoto, Y. *J. Am. Chem. Soc.* **2003**, *125*, 14133. (c) Taggi, A. E.; Hafez, A. M.; Lectka, T. *Acc. Chem. Res.* **2003**, *36*, 10. (d) Hamada, T.; Manabe, K.; Kobayashi, S. *Angew. Chem., Int. Ed.* **2003**, *42*, 3927. (e) Fernandes, R. A.; Yamamoto, Y. *J. Org. Chem.* **2004**, *69*, 735. (f) Ding, H.; Friestad, G. K. *Synthesis* **2004**, 2216. (g) Wada, R.; Shibuguchi, T.; Makino, S.; Osaki, K.; Kanai, M.; Shibasaki, M. *J. Am. Chem. Soc.* **2006**, *128*, 7687.
- (a) Berger, R.; Rabbat, P. M. A.; Leighton, J. L. *J. Am. Chem. Soc.* **2003**, *125*, 9596. (b) Berger, R.; Duff, K.; Leighton, J. L. *J. Am. Chem. Soc.* **2004**, *126*, 5686.
- (a) Kobayashi, S.; Ogawa, C.; Konishi, H.; Sugiura, M. *J. Am. Chem. Soc.* **2003**, *125*, 6610. (b) Fernández, I.; Valdivia, V.; Gori, B.; Alcudia, F.; Alvarez, E.; Khiar, N. *Org. Lett.* **2005**, *7*, 1307. (c) Ogawa, C.; Sugiura, M.; Kobayashi, S. *Angew. Chem., Int. Ed.* **2004**, *43*, 6491.
- For indium-mediated allylation of imines, see: (a) Beuchet, P.; Le Marrec, N.; Mosset, P. *Tetrahedron Lett.* **1992**, *33*, 5959. (b) Chan, T. H.; Lu, W. *Tetrahedron Lett.* **1998**, *39*, 8605. (c) Lu, W.; Chan, T. H. *J. Org. Chem.* **2000**, *65*, 8589. (d) Kumar, H. M. S.; Anjaneyulu, S.; Reddy, E. J.; Yadav, J. S. *Tetrahedron Lett.* **2000**, *41*, 9311. (e) Lu, W.; Chan, T. H. *J. Org. Chem.* **2001**, *66*, 3467. See also: (f) Prajapati, D.; Laskar, D. D.; Gogoi, B. J.; Devi, G. *Tetrahedron Lett.* **2003**, *44*, 6755. (g) Banik, B. K.; Ghatak, A.; Becker, F. F. *J. Chem. Soc., Perkin Trans. 1* **2000**, 2179. (h) Ghatak, A.; Becker, F. F.; Banik, B. K. *Heterocycles* **2000**, *53*, 2769. (i) Skaaranderup, P. R.; Madsen, R. J. *Org. Chem.* **2003**, *68*, 2115.
- (a) Cooper, I. R.; Grigg, R.; MacLachlan, W. S.; Thornton-Pett, M.; Sridharan, V. *Chem. Commun.* **2002**, 1372. (b) Cooper, I. R.; Grigg, R.; Hardie, M. J.; MacLachlan, W. S.; Sridharan, V.; Thomas, W. A. *Tetrahedron Lett.* **2003**, *44*, 2283.
- (a) Basile, T.; Bocoum, A.; Savoia, D.; Umani-Ronchi, A. *J. Org. Chem.* **1994**, *59*, 7766. (b) Loh, T.-P.; Ho, D. S.-C.; Xu, K.-C.; Sim, K.-Y. *Tetrahedron Lett.* **1997**, *38*, 865. (c) Vilaivan, T.; Winotapan, C.; Banphachit, V.; Shinada, T.; Ohfune, Y. *J. Org. Chem.* **2005**, *70*, 3464.
- (a) Lee, J. G.; Choi, K. I.; Pae, A. N.; Koh, H. Y.; Kang, Y.; Cho, Y. S. *J. Chem. Soc., Perkin Trans. 1* **2002**, 1314. (b) Miyabe, H.; Nishimura, A.; Ueda, M.; Naito, T. *Chem. Commun.* **2002**, 1454. (c) Miyabe, H.; Yamaoka, Y.; Naito, T.; Takemoto, Y. *J. Org. Chem.* **2003**, *68*, 6745. (d) Miyabe, H.; Yamaoka, Y.; Naito, T.; Takemoto, Y. *J. Org. Chem.* **2004**, *69*, 1415.
- Cook, G. R.; Maity, B.; Kargbo, R. *Org. Lett.* **2004**, *6*, 1741.
- For reviews of diastereoselective allylation of imines using chiral auxiliaries, see: (a) Lavaro, G.; Savoia, D.; *Synlett* **2002**, 651. (b) Ding, H.; Friestad, G. K. *Synthesis* **2005**, 2815.
- Cook, G. R.; Kargbo, R.; Maity, B. *Org. Lett.* **2005**, *7*, 2767.
- For enantioselective In-mediated allylation of aldehydes using stoichiometric chiral additives, see: (a) Loh, T.-P.; Zhou, J.-R. *Tetrahedron Lett.* **1999**, *40*, 9115. (b) Loh, T.-P.; Zhou, J.-R.; Li, X.-R. *Tetrahedron Lett.* **1999**, *40*, 9333. (c) Loh, T.-P.; Zhou, J.-R.; Yin, Z. *Org. Lett.* **1999**, *1*, 1855. (d) Loh, T.-P.; Lin, M.-J.; Tan, K.-L. *Tetrahedron Lett.* **2003**, *44*, 507. (e) Hirayama, L. C.; Gamsey, S.; Kneppel, D.; Steiner, D.; DeLaTorre, K.; Singaram, B. *Tetrahedron Lett.* **2005**, *46*, 2315. (f) using allylstannanes: (BINOL/InCl₃) Teo, Y.-C.; Tan, K.-T.; Loh, T.-P. *Chem. Commun.* **2005**, 1318. (g) (In(OTf)₃/PYBOX) Lu, J.; Ji, S.-J.; Teo, Y.-C.; Loh, T.-P. *Org. Lett.* **2005**, *7*, 159.
- Tan, K. L.; Jacobsen, E. N. *Angew. Chem., Int. Ed.* **2007**, *46*, 1315.
- (a) Chan, T. H.; Yang, Y. *J. Am. Chem. Soc.* **1999**, *121*, 3228. (b) Araki, S.; Ito, H.; Katsumura, N.; Butsugan, Y.; *J. Organomet. Chem.* **1989**, *369*, 291.
- Hansch, C.; Leo, A.; Taft, R. W. *Chem. Rev.* **1991**, *91*, 165.
- (a) Charmant, J. P. H.; Dyke, A. M.; Lloyd-Jones, G. C. *Chem. Commun.* **2003**, 380. (b) Zhao, Z.; Messinger, J.; Schön, U.; Wartchow, R.; Butenschön, H. *Chem. Commun.* **2006**, 3007.
- Steenisma, R. W.; Galabi, S.; Tagat, J. R.; McCombie, S. W. *Tetrahedron Lett.* **2001**, *42*, 2281.
- Kozlowski has reported on arylsulfone-2-naphthol homocoupling to yield R^H = Ar analogues of **5**, in 45–75% ee (98% ee after titration): Li, X.; Hewgley, B.; Mulrooney, C. A.; Yang, J.; Kozlowski, M. C. *J. Org. Chem.* **2003**, *68*, 5500.
- For leading references see: (a) *The Handbook of Fluorous Chemistry*; Gladysz, J. A.; Curran, D. P.; Horvath, I. T., Eds.; Wiley-VCH: Weinheim, Germany, 2004. (b) Pozzi, G.; Shepperson, I. *Coord. Chem. Rev.* **2003**, *242*, 115.

JA070742T